The City School

North Nazimabad Boys Campus



E-Worksheet

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Q-1: l	Fill in the blanks with suitable answers.
1.	Electrolysis is a process in which is passed through water to break it into its
	constituent elements.
2.	While cells are the building blocks of living things, are the building
	blocks of all matter, including living and non-living things.
3.	Cells themselves are made up of different or materials containing elements.
4.	Currently there are elements, of which are found naturally and the rest are man made.
5.	Stars are made up of two most abundant elements in the universe, and
6.	The most abundant element in our universe is
7.	The most abundant element by mass in the Earth's crust and in the human body is .
8.	Some elements are named after the scientists who discovered them for e.g. Einsteinium and e.tc.
9.	A chemical reaction is a process in which substances are formed.
10.	The physical properties of elements are their; appearance, state, colour, density, boiling and melting points, conductivity.
11.	The Russian scientist Dmitri Ivanovich (1834-1907) was the first to arrange
	the known elements during his time, in the Periodic Table.
12.	The vertical columns in the Periodic Table are called and the horizontal rows are called
13.	Going across a period from left to right, elements change from being
	toin character.
14.	Liquid nitrogen is used as a agent.
15.	is a metal which is not solid at room temperature, it is semi solid/liquid.
16.	A non metal conducts electricity well like metals.
Q.2-Enc	circle the best answer from the given options.
1.Th	ne elements have same properties in
•	A period
•	A group
	ne elements that have properties of metals and non metals both are called
	Metalloids
•	Metallic elements

3. Solid carbon dioxide is called

4.A compound is a/an

- Impure substance
- Pure substance
- 5. Burning, heating and electrolysis are examples of
 - Physical change
 - Chemical reaction

6.Distilled water is

- A compound
- A mixture
- 7. Muddy water is an example of
 - compound
 - mixture

8.Mixtures are

- Pure substances
- Impure substances